



# New Analytical Tools for the Study of Bioactive Polyphenols in *Prunus avium* L. (sweet cherry) and Evaluation of their Anti-oxidant Activity on an *In vivo* Model

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# **Summary**

Background and aim

- 2. Analytical characterisation
  - Conventional analytical techniques (HPLC-UV-ESI-MS and MS<sup>2</sup>)
  - Electrochemical analysis (sensors)

3. In vivo assessment of the anti-oxidant activity on C. elegans model

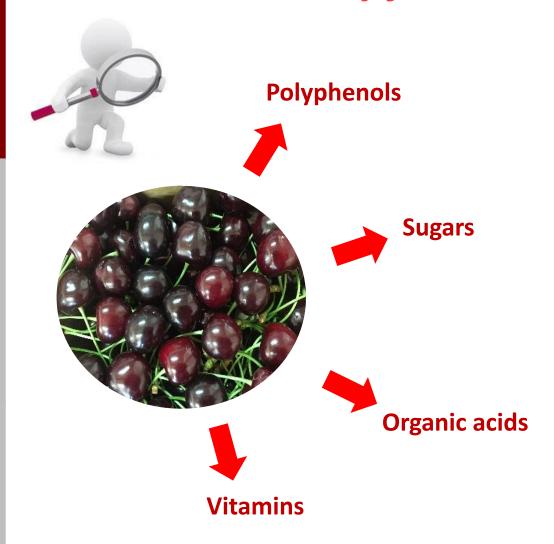
In collaboration with the research group of Prof. Celestino Santos-Buelga of the University of Salamanca

#### Prunus avium L. I.D.

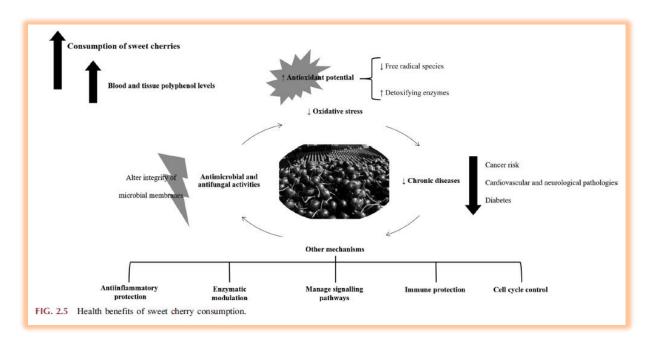


- Family: Rosaceae
- Genus: Prunus
- Species: avium
- Nickname: sweet cherry
- Fruit: drupe

# Inside sweet cherry fruit



# «Let food be your medicine and medicine be your food»- Hippocrates



*Gonçalves et al.*, Sweet Cherry Phenolic Compounds: Identification, Characterization and Health Benefits, in: Atta-ur-Rahman (Ed.), Studies in Natural Products Chemistry, Elsevier Inc., Amserdam, 2018.

# Our goal

• Application of suitable analytical techniques to the **characterization** of sweet cherry polyphenolic fraction **and development of a fast and innovative tool** alternative to conventional techniques.

• Evaluation of the capacity of sweet cherry extract to increase the **resistance to thermal stress** in *C.elegans in vivo* model.



# Sample



In this work, cherries belonging to the variety «**Moretta**» were considered. The sample was harvested near the city of Vignola in June 2018 and kept at -20 °C until the analysis.



# **Highlights**

1. Background and aim

- 2. Analytical characterisation
  - Conventional analytical techniques (HPLC-UV-ESI-MS and MS<sup>2</sup>)
  - Electrochemical analysis (sensors)

3. In vivo assessment of the anti-oxidant activity on C. elegans model

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# Workflow

Extraction of secondary metabolites

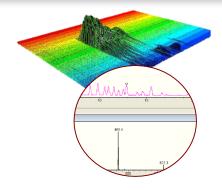
Quantification of the main classes of polyphenols (Uv-Vis)

Identification of the compounds (HPLC-UV/DAD and MS and MS<sup>2</sup>)

Electrochemical analysis

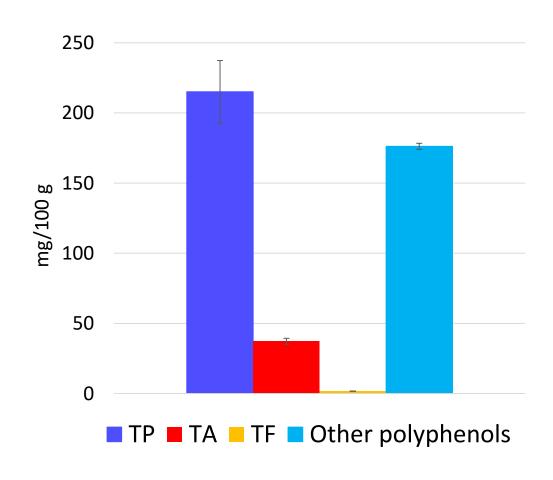


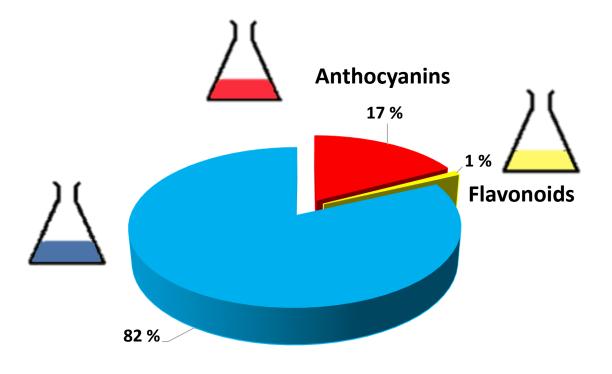






# Spectrophotometric analyses





Other phenolics

Which anthocyanins?
Which flavonoids?
Which other phenolics?



# **Compounds identification**

HPLC-ESI-UV/DAD and MS and MS<sup>2</sup> analysis: combining spectral, MS and fragmentation information.

#### **Chromatographic parameters**

Chromatographic apparatus: HPLC Agilent Technologies modular model 1100;

**Column tested:** Zorbax  $C_{18}$  SB (150 x 4.6 mm, 5  $\mu$ m, Agilent Technologies),

Mobile phase:

**A)**  $H_2O$ -HCOOH (9:1, v/v),

**B**) MeOH-H<sub>2</sub>O-HCOOH (5:4:1, v/v/v), gradient elution;

**Injection volume:** 5 μL;

**Flow:** 1.0 mL/min (for MS and MS<sup>2</sup> analysis flow was splitted 1:5);

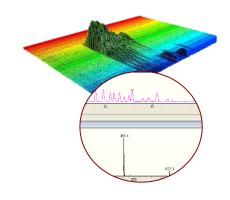
**Column temperature:** 25°C;

#### **Detection parameters**

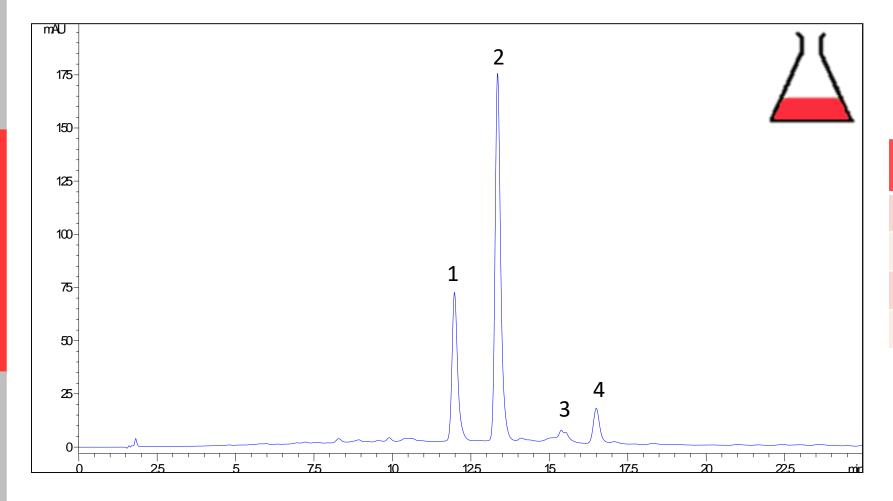
**UV/DAD detection:** 280 nm and 320 nm (phenolic acids and flavonoids), 520 nm (anthocyanins);

**MS-IT detection:** The ion trap mass analyzer was used in the full-scan **positive** (anthocyanins) and **negative** ion modes (flavonoids, phenolic acid), in the m/z range 100-1000. MS<sup>2</sup> spectra were automatically performed with helium as the collision gas, using the SmartFrag function.





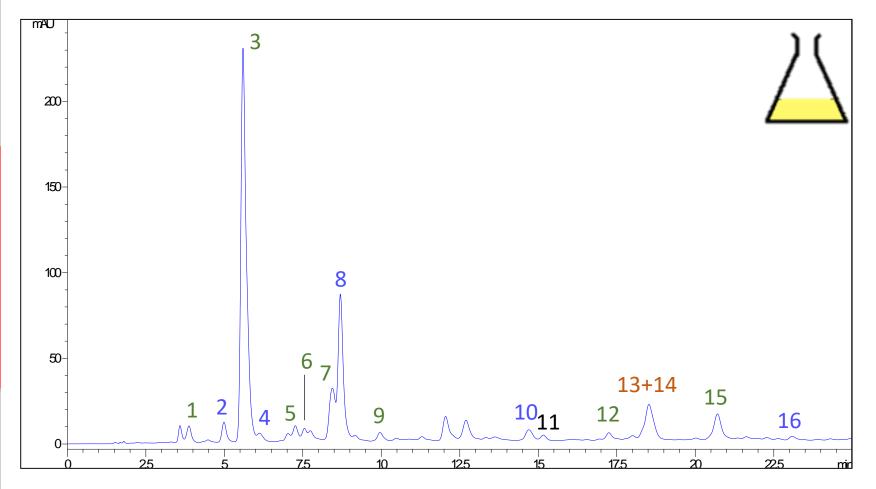
# **Compounds identification**



| Peak<br>n° | Anthocyanin                 |
|------------|-----------------------------|
| 1          | Cyanidin-3-O-glucoside      |
| 2          | Cyanidin-3-O-rutinoside     |
| 3          | Pelargonidin-3-O-rutinoside |
| 4          | Peonidin-3-O-rutinoside     |

Representative HPLC-UV/DAD chromatogram of **sweet cherry fruit extract** recorded at 520 nm

# **Compounds identification**



| Representative HPLC-UV/DAD | chromatogram of <b>sweet</b> | cherry fruit |
|----------------------------|------------------------------|--------------|
| extract recorded at 280 nm |                              |              |

Hydroxycinnamic acids

Flavan-3-ols

Flavonols

Other Flavonoids

| Peak<br>n° | Compound                          |
|------------|-----------------------------------|
| 1          | Neochlorogenic acid               |
| 2          | Catechin                          |
| 3          | 3-Coumaroylquinic acid            |
| 4          | Procyanidin dimer B type          |
| 5          | 5-Caffeoylquinic acid             |
| 6          | Ferruoyilquinic acid              |
| 7          | cis-4-Coumaroylquinic acid        |
| 8          | Epicatechin                       |
| 9          | trans-4-Coumaroylquinic acid      |
| 10         | Procyanidin dimer B type          |
| 11         | Naringenin-hexoside               |
| 12         | 3,5-Dicaffeoylquinic acid         |
| 13         | Quercetin-3-O-glucoside           |
| 14         | Quercetin-3-O-rutinoside          |
| 15         | 3-Coumaroyl-5-caffeoylquinic acid |
| 16         | Catechin-hexoside                 |

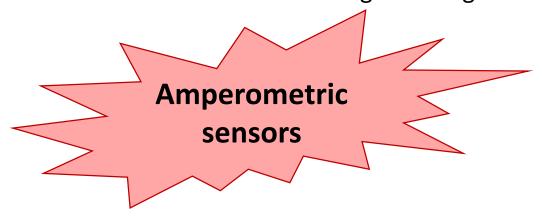
#### Come to the unconventional side!

#### **Conventional techniques**

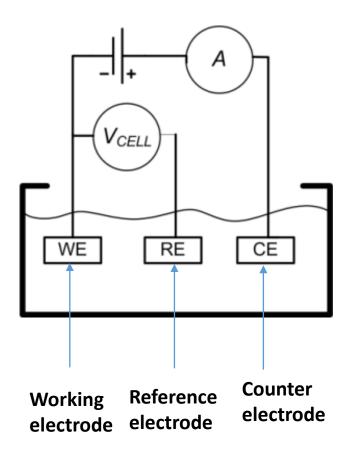
- High costs
- Time consuming sample pre-treatment
- Large sample size
- Large solvents and reagents usage

#### **Electrochemical analysis**

- Low costs
- Fast (minimal sample pre-treatment)
- Small sample size
- Limited solvents and reagents usage



# **Amperometric sensors**



**Figure.** Scheme of a typical circuit used with amperometric sensors.

- Chemical sensors
- measure current response to detect the concentration of an analyte at a fixed potential.
- The current is generated by a **RedOx process** between the sensor surface and the analyte.
- RedOX reaction occurs at the working electrode, (the real sensor) and determines the passage of current which, in particular conditions, is proportional to the analyte concentration.



#### **Electrochemical conditions**

- Working electrode: Sonogel Carbon electrode containing Carbon Black (SNGC\_CB);
- Reference electrode: saturated Ag/AgCl;
- Auxiliary electrode: glassy carbon;
- Working solution: 0.1 M tartrate buffer solution (TBS), pH= 3.5;

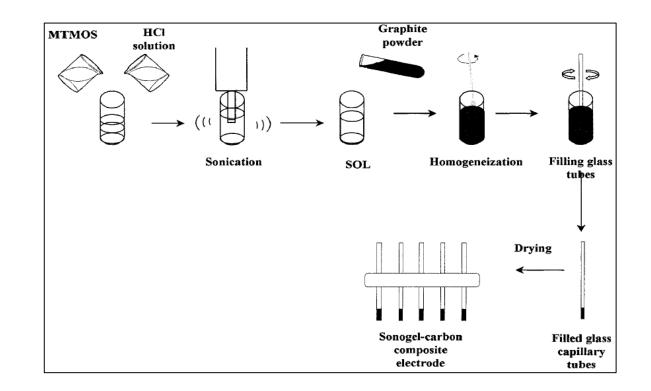
Differential Pulse Voltammetry (DPV) parameters:

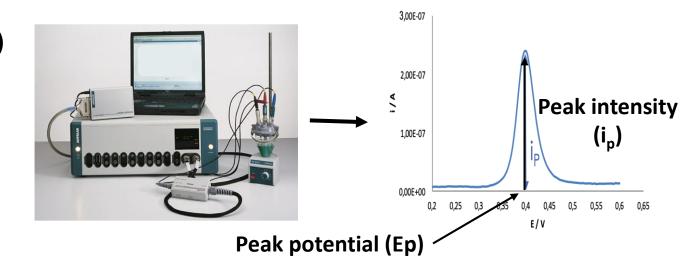
Pulse time: 0.15 s

Voltage steptime: 0.6 s

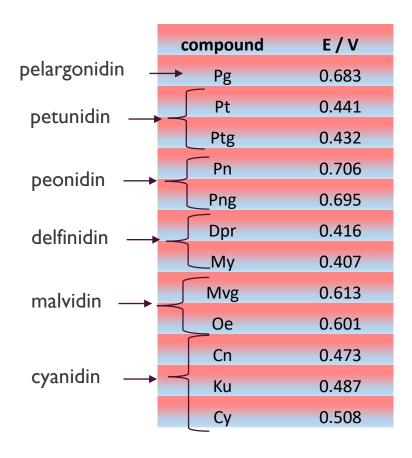
Modulation amplitude: 0.01 V

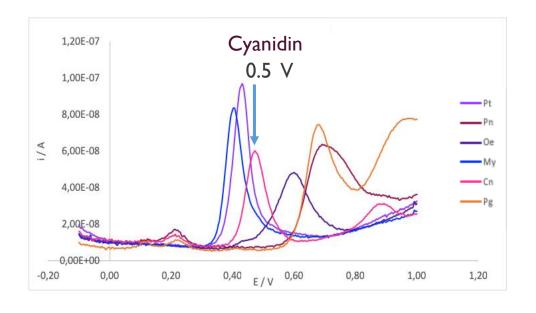
Equilibration time: 5 s



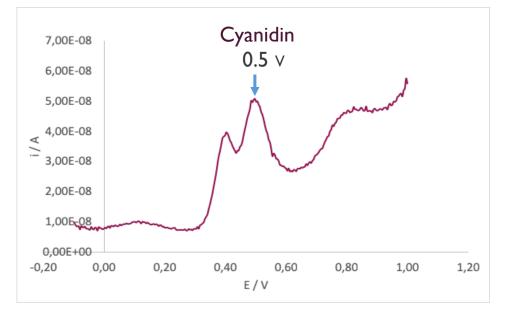


# Qualitative analysis of anthocyanins





Anthocyanins reference standard solution



Mashed sweet cherry fruit

# **Summary**

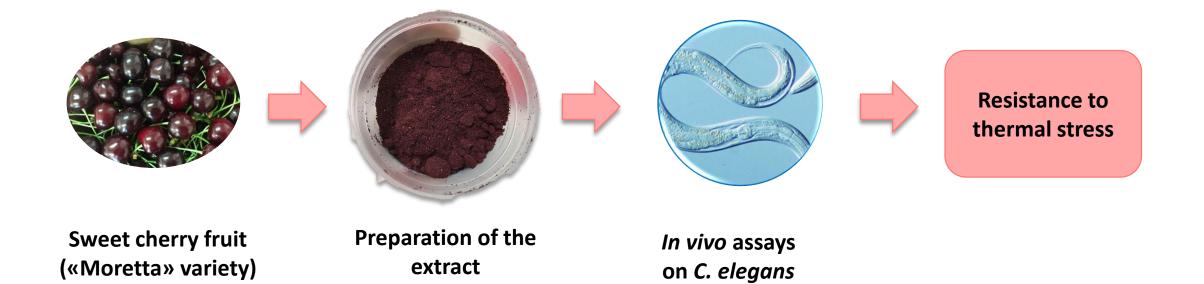
1. Background and aim

2. Analytical characterisation of sweet-cherry fruit extracts

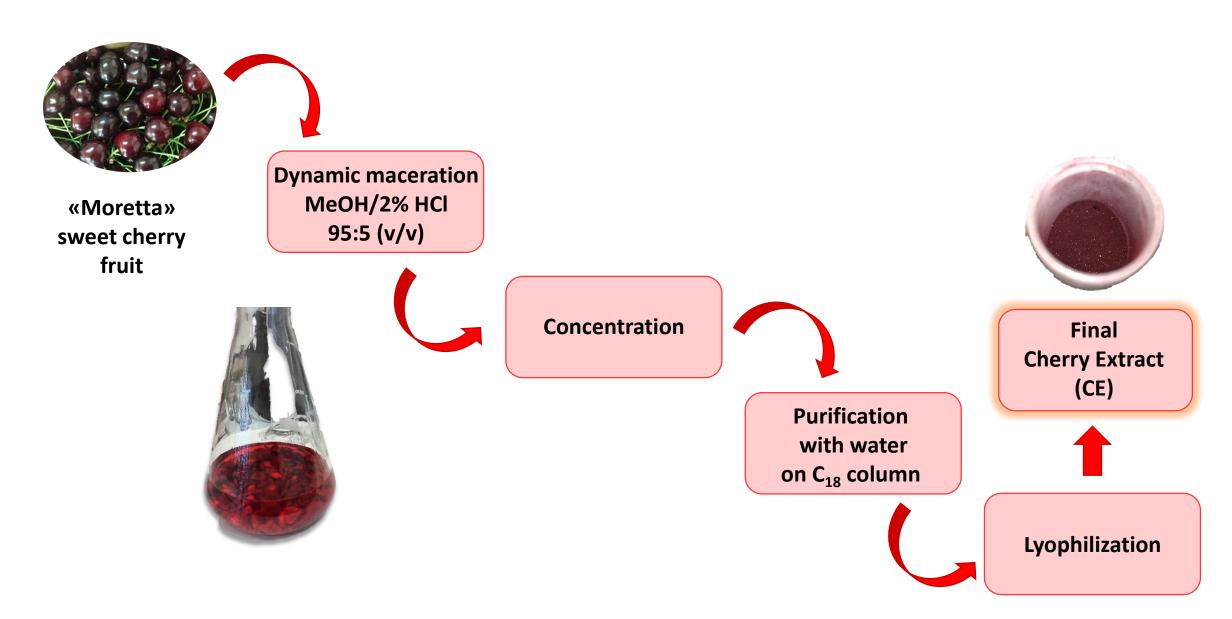
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# Workflow

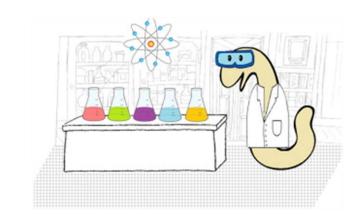


# **Preparation of the extract**



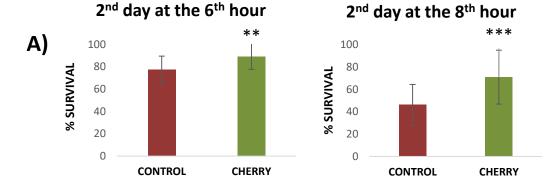
# **Experimental**

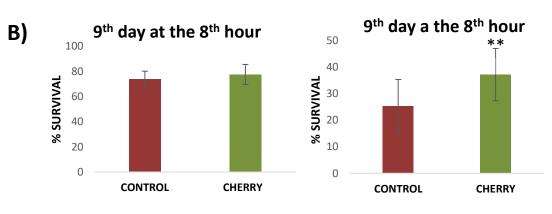
- The effects on the **resistance to thermal stress** of the extract obtained from "Moretta" cherries have been evaluated.
- Thermal stress assays were carried out in **wild type N2** of the model nematode *C. elegans* **grown** on NGM agar plates **in the presence and absence** (control group) of the **cherry extract** at different concentrations, on the basis of previous experiments.
- Resistance to thermal stress was evaluated by measuring the **rate of survival**, through the response to a mechanical stimulus, after **6 or 8 hours** of **exposure** to the **lethal temperature of 35 °C** at different stages of the development (2<sup>nd</sup> and 9<sup>th</sup> day of adulthood).
- Assays were performed with approximately 100 nematodes for treatment, and each trial was made in triplicate. In each assay, the results were expressed as percentage of living animals.



#### Results

- None of the concentration of the cherry extract tested leaded to toxic effect on nematodes.
- The most significant results on the anti-oxidant activity were obtained at a concentration of 25 μg/mL. Indeed, an **increase in survival** in the nematodes treated with cherry extract both on the **second and ninth day** of worm life was evident.
- The most significant result was obtained on the second day (A) for a thermal stress of eight hours where there was a 25% increase in the survival of treated worms compared to the control. The differences were considered significant at \*(p<0,05), \*\*(p<0,01) and \*\*\*(p<0,001).</p>





#### **Overall conclusions**

- In this work, cherry fruit belonging to the «Moretta» variety were characterized in the composition of their polyphenolic fraction by means of conventional analytical techniques. The **polyphenolic fraction** of «Moretta» cherry fruit encompassed **anthocyanins**, **hydroxycinnamic acids**, **flavan-3-ols** and other flavonoids.
- An innovative analytical tool based on electrochemical analysis with SNGC\_CB amperometric sensors for the
  fast determination of anthocyanins in sweet cherry fruit was developed. The sensor proved capable of
  selectively identify the major class of anthocyanins in cherry fruit, namely cyanidin-3-O-glycosides. Future steps
  will include the quantitative determination of the anthocyanins by using this innovative technique.
- Given the high biological value of the polyphenolic fraction of "Moretta" cherry fruit, its extract was submitted to *in vivo* antioxidant activity assays by using the *C. elegans* model. The extract was able **to significantly increase the adult worm resistance to oxidative and thermal stress** at a concentration of **25 μg/mL**.
- Moretta cherry extract represents a promising source of bioactive compounds, thus it might find an application
  in the nutraceutical industry.

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# Thank you!!

...questions?



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